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Key indicators

Single-crystal X-ray study T = 294 KMean σ (C–C) = 0.015 Å R factor = 0.065 wR factor = 0.194 Data-to-parameter ratio = 16.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

1-(2,2-Dibromovinyl)-3,5-dimethoxybenzene

In the title compound, $C_{10}H_{10}Br_2O_2$, the angle between the dibromovinyl group and the benzene ring is $42.2 (2)^{\circ}$.

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Comment

The title compound, (I), is a vinylic dibromide derivative prepared from 3,5-dimethoxybenzaldehyde (Papahatjis et al., 1998) and its structure is presented here (Fig. 1).



In styrene, both the benzene ring and the double bond are coplanar due to conjugation. This is not possible in the case of (I) due to steric repulsion between the vinylic Br atom and benzene CH moieties. The double bond and the benzene ring of the title compound are inclinhed at a dihedral angle of 42.2 (2)°.

Experimental

The title compound was prepared according to the procedure of Papahatjis et al. (1998). Suitable crystals were obtained by evaporation of an ethyl acetate/hexane (1:4 v/v) solution (m.p. 345 K).

Crystal data	
$C_{10}H_{10}Br_{2}O_{2}$ $M_{r} = 321.98$ Monoclinic, P_{21}/n a = 4.129 (2) Å b = 11.192 (6) Å c = 25.722 (15) Å	Z = 4 $D_x = 1.800 \text{ Mg m}^{-3}$ Mo K\alpha radiation $\mu = 6.80 \text{ mm}^{-1}$ T = 294 (2) K Plack exclusion
$\begin{aligned} c &= 25.752 (15) \text{ A} \\ \beta &= 92.274 (10)^{\circ} \\ V &= 1188.2 (11) \text{ Å}^3 \end{aligned}$	$0.26 \times 0.24 \times 0.12 \text{ mm}$
Bruker APEX-II CCD area- detector diffractometer φ and ω scans Absorption correction: multi-scan (<i>SADABS</i> ; Sheldrick, 1996) $T_{\min} = 0.186, T_{\max} = 0.442$	5582 measured reflections 2086 independent reflections 1422 reflections with $I > 2\sigma(R_{int} = 0.053)$ $\theta_{max} = 25.0^{\circ}$

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 $l > 2\sigma(I)$

Refinement

Refinement on F^2 $w = 1/[\sigma^2(F_0^2) + (0.0831P)^2]$ $R[F^2 > 2\sigma(F^2)] = 0.066$ wR(F²) = 0.194 S = 1.182086 reflections 130 parameters H-atom parameters constrained

+ 5.9795P] where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 1.00 \text{ e} \text{ Å}^{-3}$ $\Delta \rho_{\rm min} = -0.82 \text{ e} \text{ Å}^{-3}$ Extinction correction: SHELXL Extinction coefficient: 0.013 (2)

All H atoms were positioned geometrically and refined as riding (C-H = 0.93-0.96 Å); for the CH and CH₂ groups, $U_{iso}(H)$ values were set equal to $1.2U_{eq}(C)$, and $1.5U_{eq}(C)$ for the methyl groups.

Data collection: CrystalClear (Molecular Structure Corporation & Rigaku, 1999); cell refinement: CrystalClear; data reduction: CrystalStructure (Molecular Structure Corporation & Rigaku, 2005); program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997): software used to prepare material for publication: CrystalStructure.

References

Bruker (1997). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.

Molecular Structure Corporation & Rigaku (1999). CrystalClear. Version 1.3.6. MSC, The Woodlands, Texas, USA, and Rigaku Corporation, Tokyo, Japan.



Figure 1

The molecular structure of (I), with the atom numbering scheme, showing displacement ellipsoids drawn at the 35% probability level.

- Molecular Structure Corporation & Rigaku (2005). CrystalStructure. Version 3.7.0. MSC, The Woodlands, Texas, USA, and Rigaku Corporation, Tokyo, Japan.
- Papahatjis, D. P., Kourouli, T., Abadji, V., Goutopoulus, A. & Makriyannis, A. (1998). J. Med. Chem. 41, 1195-1200.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.